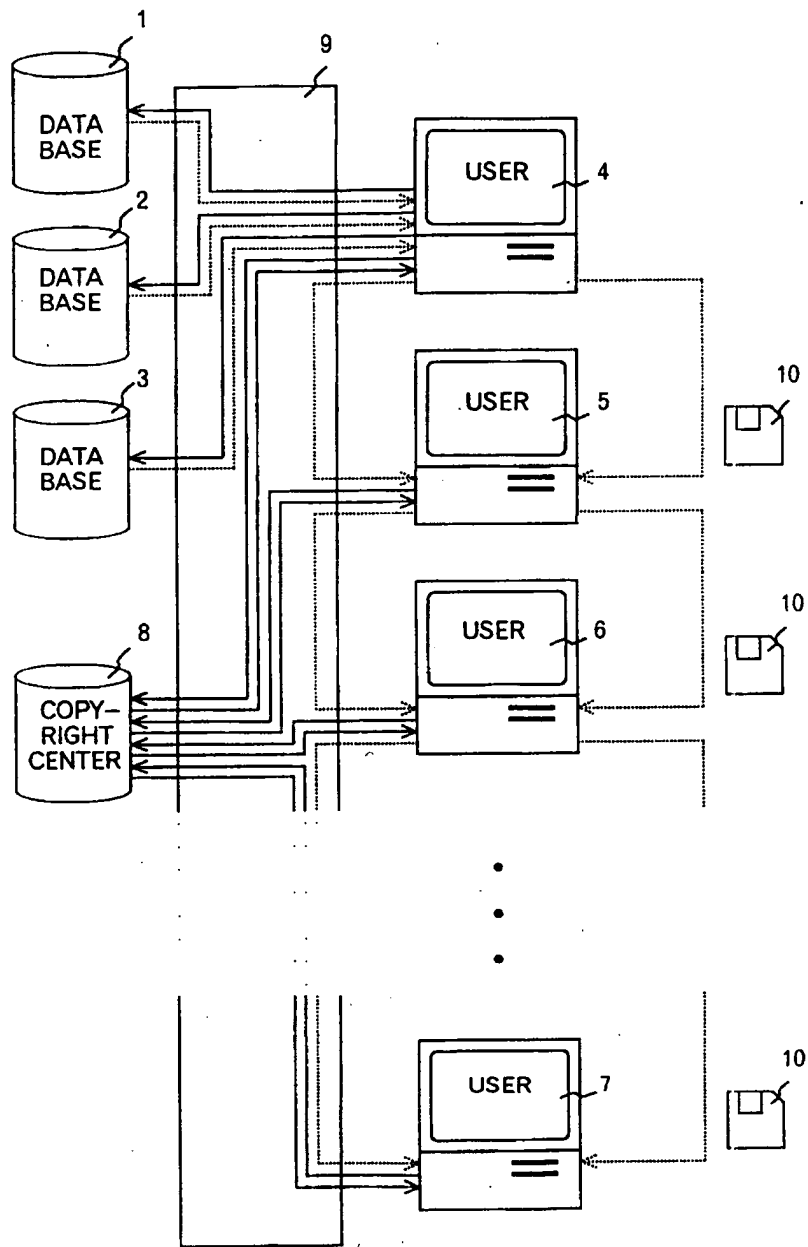


Fig. 1



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Fig. 2

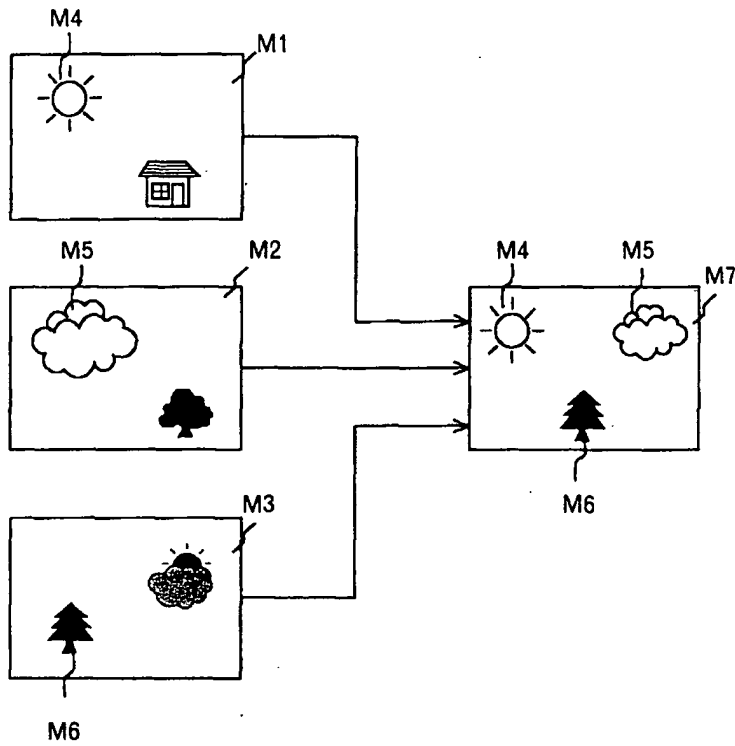


Figure 6. The effect of the initial concentration of the monomer (C_0) on the apparent rate constant (k_{app}) at different temperatures. The experimental conditions were: $[C_0] = 0.08 \text{ mol/L}$, $[I] = 0.001 \text{ mol/L}$, $[AIBN] = 0.001 \text{ mol/L}$, $[K_2S_2O_8] = 0.001 \text{ mol/L}$, $[NaHCO_3] = 0.001 \text{ mol/L}$, $[NaHSO_3] = 0.001 \text{ mol/L}$, $[NaOH] = 0.001 \text{ mol/L}$, $[NaCl] = 0.001 \text{ mol/L}$, $[NaNO_3] = 0.001 \text{ mol/L}$, $[Na_2SO_4] = 0.001 \text{ mol/L}$, $[Na_2CO_3] = 0.001 \text{ mol/L}$, $[Na_2HPO_4] = 0.001 \text{ mol/L}$, $[Na_2HPO_4] = 0.001 \text{ mol/L}$, $[Na_2HPO_4] = 0.001 \text{ mol/L}$, $[Na_2HPO_4] = 0.001 \text{ mol/L}$.

Fig. 3

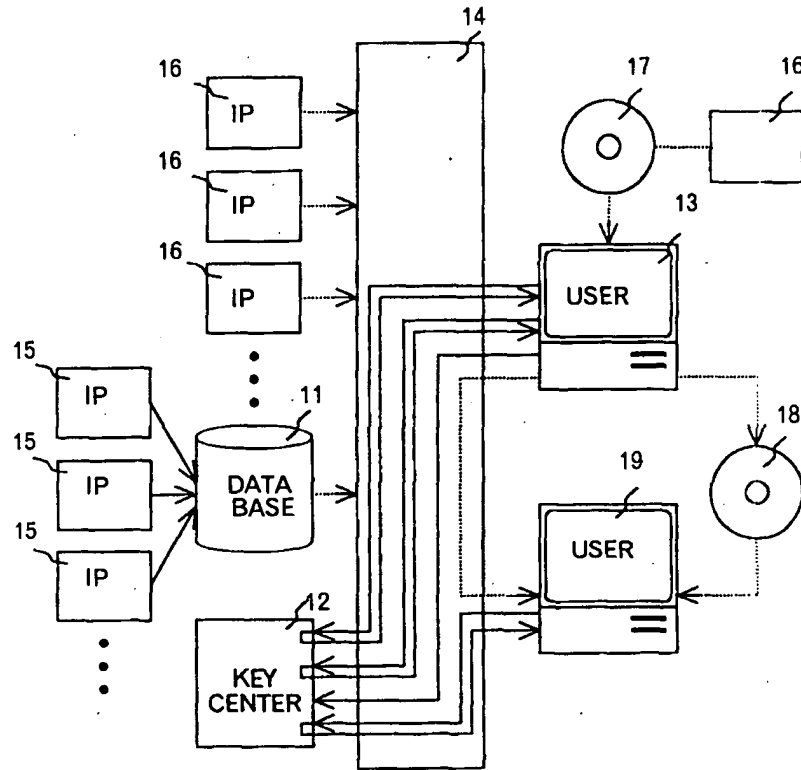


Fig. 4

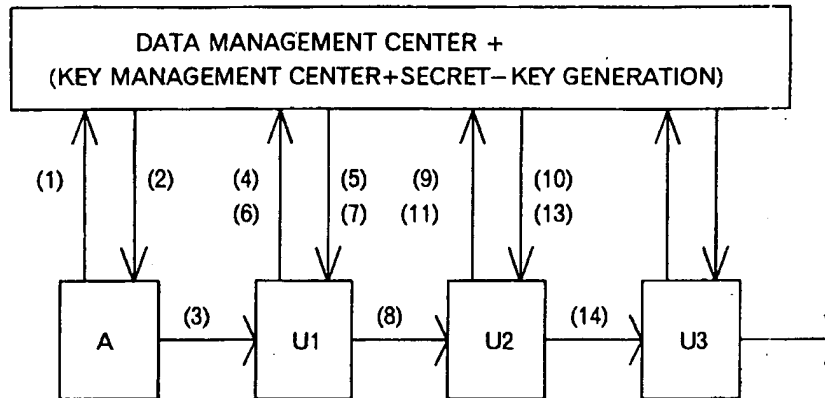


Fig. 5

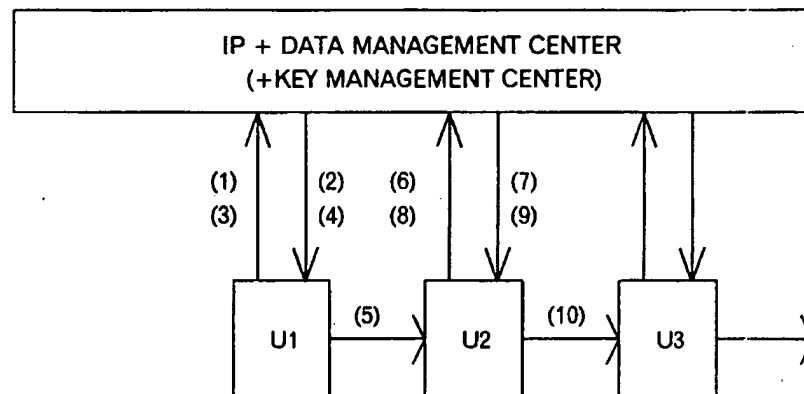


Fig. 6

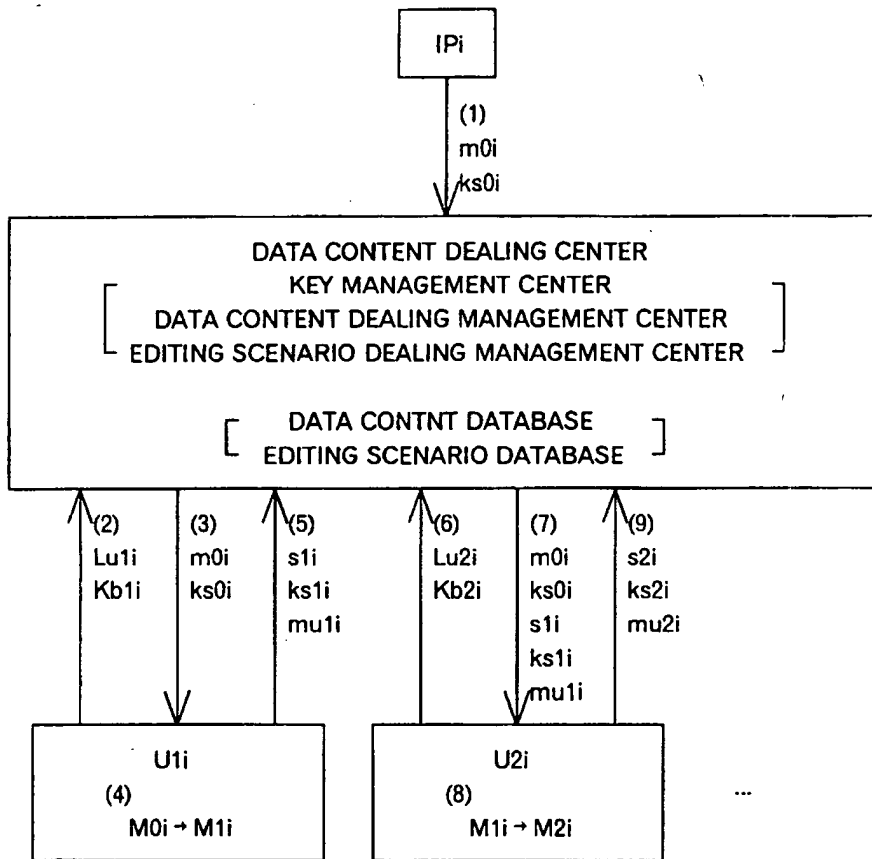
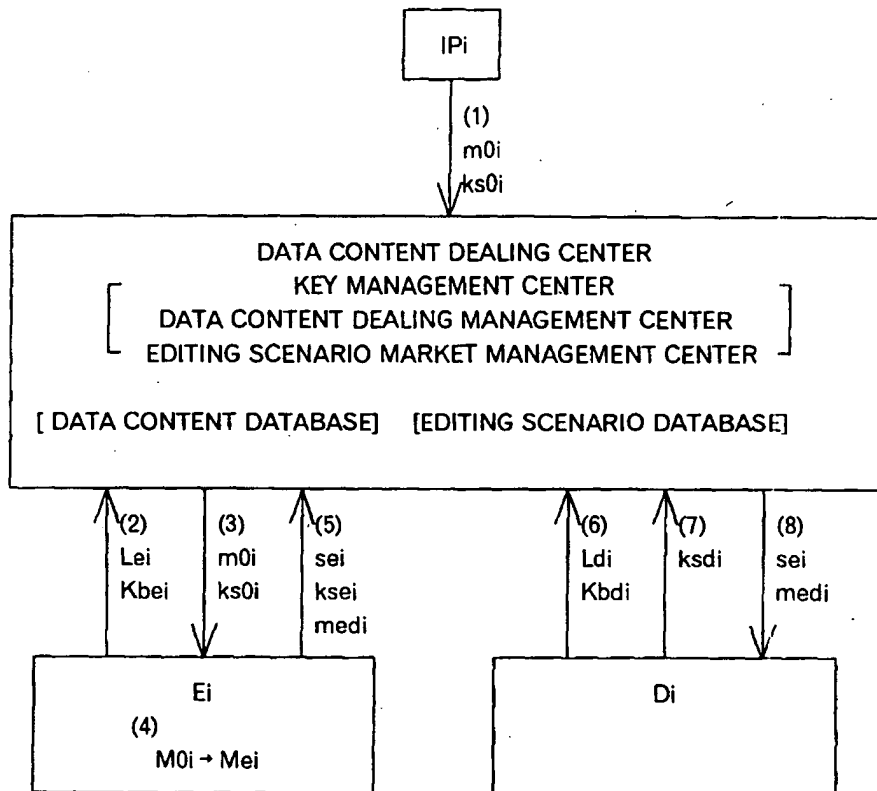


Fig. 7



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